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Roll No.

Total No. of Questions: 09]

[Total No. of Pages: 02

B.Tech. (Sem. - 1st/2nd) ENGINEERING CHEMISTRY

SUBJECT CODE: CH - 101 (2K4 & Onwards)

<u>Paper ID</u>: [A0112]

[Note: Please fill subject code and paper ID on OMR]

Time: 03 Hours

Maximum Marks: 60

Instruction to Candidates:

- 1) Section A is Compulsory.
- 2) Attempt any Five questions from Section B & C.
- 3) Select atleast Two questions from Section B & C.

Section - A

Q1)

(Marks: 2 Each)

- a) Cause of chemical shift in NMR.
- b) Specifications for water to be used for drinking purpose.
- c) What is priming and foaming.
- d) Difference between Galvanic cell and Electrolytic cell.
- e) Corrosion is reverse to extraction. Justify.
- f) What is meant by metastable equilibrium?
- g) Photosensitized reactions?
- h) Difference between GC and HPLC technique of chromatography.
- i) Cause of alkalinity of saw water.
- j) Advantages and limitations of lime soda process.

Section - B

(Marks: 8 Each)

- Q2) (a) Describe method involved in treatment of water required for industries.
 - (b) State any two difference between hot and cold lime soda process.

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- 03) (a) What is soil corrosion? What are the factors that affects soil corrosions.
 - (b) Explain mechanism of rusting of Iron in acidic and neutral environment.
- Q4) (a) Describe various kinds of supports used in columns and their properties.
 - (b) Give schematic diagram of HPLC apparatus showing various parts.
- Q5) (a) Define emf. How is the emf of a cell determined?
 - (b) Derive an expression for the emf of a concentration cell.

Section - C

(Marks: 8 Each)

- **Q6)** (a) What do you understand by term spin-spin coupling? Is the coupling constant independent of the applied field or depend on it.
 - (b) Indicate diagrammatically the splitting of signals in NMR spectra of CH₂Br-CHBr₂, CH₃-CHBR₂ and CH₃-CH₂Br.
- **Q7)** (a) What type of molecules give pure rotational spectra? Derive expression for frequency of rotational lines.
 - (b) What type of molecules exhibit vibrational-rotational spectra and why?
- Q8) (a) How does CO₂ system differ from water system?
 - (b) Explain the terms Eutectic point, Eutectic composition, constant boiling mixture, triple point.
- **Q9)** (a) A sample of gaseous HI was irradiated by light of wavelength 253.7 nm when 307 J of energy was found to decompose 1.30×10^{-3} moles of HI. Calculate quantum yield for the dissociation of HI.
 - (b) State and explain Einstein law of photochemical equivalence. What is meant by quantum yield of a photochemical process.

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